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Key indicators

Single-crystal X-ray study T = 298 KMean $\sigma(\text{C-C}) = 0.004 \text{ Å}$ R factor = 0.047 wR factor = 0.108Data-to-parameter ratio = 17.4

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

Bis[2-(cyclohexyliminomethyl)-4-nitrophenolato]zinc(II)

In the mononuclear title compound, $[Zn(C_{13}H_{15}N_2O_3)_2]$, the Zn^{II} atom is four-coordinated in a distorted tetrahedral configuration by two imine N and two phenolate O atoms from two Schiff base ligands.

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Comment

Previously, some Schiff base zinc(II) compounds (You, 2005a,b,c) have been reported. As an extension of the work on these compounds, the title zinc(II) compound, (I), is reported here.

$$O_2N$$
 O_2N
 O_2N

Complex (I), a mononuclear zinc(II) compound (Fig. 1), is structurally similar to bis[2-(cyclopropyliminomethyl)phenolato]zinc(II), [(II); You *et al.*, 2003]. The bond lengths and angles (Table 1) in (I) are comparable to those in (II). The central $\mathbf{Zn^{II}}$ atom is four-coordinated by two imine N and two phenolate O atoms from two Schiff base ligands. This $\mathbf{ZnN_{2}O_{2}}$ coordination forms a distorted tetrahedral geometry, with angles subtended at the $\mathbf{Zn^{II}}$ atom in the range 96.25 (7)–123.95 (8)°. In the crystal structure, centrosymmetrically related molecules are linked through $\mathbf{C-H\cdots O}$ hydrogen bonds (Table 2) and $\mathbf{C-H\cdots \pi}$ interactions involving the C1–C6 benzene ring (centroid $\mathit{Cg1}$).

Experimental

Cyclohexylamine (0.1 mmol, 9.9 mg) and 5-nitrosalicylaldehyde (0.1 mmol, 16.7 mg) were dissolved in MeOH (10 ml). The mixture was stirred at room temperature for 10 min to give a clear yellow solution. To this solution was added a MeOH solution (5 ml) of Zn(CH₃COO)₂·4H₂O (0.1 mmol, 25.6 mg), with stirring. The resulting mixture was stirred for another 10 min at room temperature. After keeping the filtrate in air for 11 d, colourless block-shaped crystals were formed at the bottom of the vessel.

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metal-organic papers

Crystal data

$[Zn(C_{13}H_{15}N_2O_3)_2]$	Z = 2
$M_r = 559.91$	$D_x = 1.443 \text{ Mg m}^{-3}$
Triclinic, $P\overline{1}$	Mo $K\alpha$ radiation
a = 10.759 (1) Å	Cell parameters from 4267
b = 11.108 (1) Å	reflections
c = 12.439 (1) Å	$\theta = 2.3 - 25.5^{\circ}$
$\alpha = 113.74 \ (1)^{\circ}$	$\mu = 1.00 \text{ mm}^{-1}$
$\beta = 104.39 \ (1)^{\circ}$	T = 298 (2) K
$\gamma = 94.99 \ (1)^{\circ}$	Block, colourless
$V = 1288.5 (2) \text{ Å}^3$	$0.21 \times 0.12 \times 0.11 \text{ mm}$

Data collection

Bruker SMART CCD area-detector diffractometer 4828 reflections with $I > 2\sigma(I)$ ω scans $R_{\rm int} = 0.034$ $\theta_{\rm max} = 27.5^{\circ}$ $h = -13 \rightarrow 13$ $T_{\rm min} = 0.817, T_{\rm max} = 0.898$ $k = -14 \rightarrow 14$ $l = -16 \rightarrow 16$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_0^2) + (0.0471P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.047$	+ 0.1392P
$wR(F^2) = 0.108$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.09	$(\Delta/\sigma)_{\text{max}} = 0.001$
5819 reflections	$\Delta \rho_{\text{max}} = 0.29 \text{ e Å}^{-3}$
334 parameters	$\Delta \rho_{\min} = -0.35 \text{ e Å}^{-3}$
H-atom parameters constrained	

 Table 1

 Selected geometric parameters (\mathring{A} , °).

Zn1-O2	1.918 (2)	Zn1-N1	1.998 (2)
Zn1-O1	1.927 (2)	Zn1-N2	2.000(2)
O2-Zn1-O1	119.44 (8)	O2-Zn1-N2	96.40 (7)
O2-Zn1-N1	112.74 (8)	O1-Zn1-N2	109.86 (8)
O1-Zn1-N1	96.25 (7)	N1-Zn1-N2	123.95 (8)

Table 2 Hydrogen-bond geometry (Å, $^{\circ}$).

D $ H$ \cdots A	D-H	$H \cdot \cdot \cdot A$	$D \cdot \cdot \cdot A$	$D-\mathbf{H}\cdot\cdot\cdot A$
$ \begin{array}{c} C8 - H8 \cdot \cdot \cdot O3^{i} \\ C24 - H24B \cdot \cdot \cdot Cg1^{ii} \end{array} $	0.98	2.43	3.380 (3)	163
	0.97	2.78	3.680 (4)	155

Symmetry codes: (i) -x + 1, -y, -z + 1; (ii) -x + 1, -y + 1, -z + 1. Note: Cg1 is the centroid of the C1–C6 benzene ring.

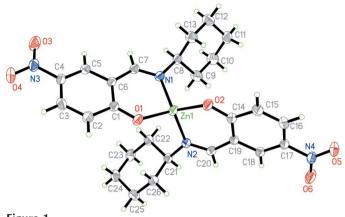


Figure 1
The structure of (I), showing the atom-numbering scheme. Displacement ellipsoids are drawn at the 30% probability level.

H atoms were placed in geometrically idealized positions and constrained to ride on their parent atoms, with C—H distances in the range 0.93–0.98 Å and $U_{\rm iso}({\rm H})=1.2U_{\rm eq}({\rm C})$.

Data collection: *SMART* (Bruker, 1998); cell refinement: *SAINT* (Bruker, 1998); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997a); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997a); molecular graphics: *SHELXTL* (Sheldrick, 1997b); software used to prepare material for publication: *SHELXTL*.

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